Part 1: Before you use the models

You will use models to learn about photosynthesis at the atomic-molecular scale, as you continue to look for answers to "unanswered questions" from your investigation.

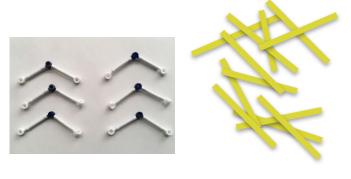
Introduction

 Plants photosynthesize when they are in the light. Light energy is changed into chemical energy, which is stored in the high-energy bonds of glucose: C-C and C-H bonds. Use the molecular models to show how this happens.

Part 2: Building the reactants

- 1. Use the bonds to make models of 6 carbon dioxide molecules (CO_2) and 6 water molecules (H_2O). You will need to use double bonds for your CO_2 .
- 2. Place some twist ties next to the molecules. These represent light energy.
- 3. Compare your molecules to the pictures below. Are they the same?





Answer Checkpoint 1 on your worksheet and check answers on wall

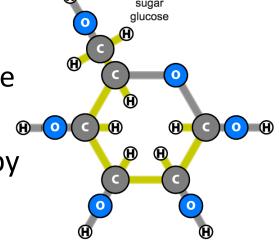
Part 3: Show the chemical change

- 1. Take the carbon dioxide and water molecules apart and recombine them into glucose $(C_6H_{12}O_2)$ and oxygen (O_2) molecules. Put these molecules on the *product* side of your table.
- 2. Identify the high-energy bonds (C-C and C-H) by putting twist ties on them. How many high energy bonds does a molecule of sugar have?
- 3. Compare your molecules to the pictures below.
 Are they the same?





Answer Checkpoint 2 on your worksheet and check answers on wall



Part 4: Clean up and questions

- Take apart your molecules they should be completely separated! Put all atoms, bonds and twist ties back in your cup
- 2. Finish the questions on your worksheet.

